



# Synthesis of *gem*-difluoromethylenated analogues of anamarine

Jing Lin<sup>a</sup>, Xiao-Long Qiu<sup>b</sup>, Feng-Ling Qing<sup>a,b,\*</sup>

<sup>a</sup> College of Chemistry, Chemical Engineering and Biotechnology, Donghua University, 2999 North Renmin Lu, Shanghai 201620, China

<sup>b</sup> Key Laboratory of Organofluorine Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Lu, Shanghai 200032, China

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## ABSTRACT

Practical synthesis of two *gem*-difluoromethylenated analogues of anamarine was described. The important synthetic steps included the preparation of the key intermediates **20–21** through the indium-mediated *gem*-difluoropropargylation of aldehyde **18** with the fluorine-containing building block **19** and efficient construction of  $\alpha,\beta$ -unsaturated- $\delta$ -lactone scaffold via BAIB/TEMPO procedure.

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## 1. Introduction

Isolated from the flowers and leaves of a Peruvian *Hyptis* species by Valverde and co-workers in 1979 [1,2], (+)-anamarine **1** has a structure featuring *R*-configuration in the 5,6-dihydro-2H-pyran-2-one ring and *L*-gluco arrangement in the C6-side chain (Fig. 1). In the past years, (+)-anamarine **1** along with other members of this  $\alpha,\beta$ -unsaturated lactone class of natural products [3,4] including spicigerolide **2**, hyptolide **3** and synrotolide **4** attracted considerably more attention from organic chemist, medicinal chemist and biological chemist in view of the fact that all of these compounds possessed a range of pharmacological properties, such as cytotoxicity against human tumor cells, antimicrobial or antifungal activities [5,6]. Thus, pharmacological properties of these types make these compounds and their analogues interesting synthetic goals.

So far, several groups have reported the synthesis of the anamarine and its *epi*-isomers based on two main synthetic routes (Scheme 1). One route involved the Wittig reaction between pre-modified phosphonium salt **5** and protected aldehyde **6** to afford the *Z*-olefin **7**, which was converted to anamarine or its *epi*-isomers via a series of transformations of protecting groups and *h* $\nu$ -irradiated isomerization of *Z*-double bond [7–10]. Another route utilized the ring-closing metathesis (RCM) reaction of triene compounds **8** to construct 5,6-dihydro-2H-pyran-2-one ring [11–

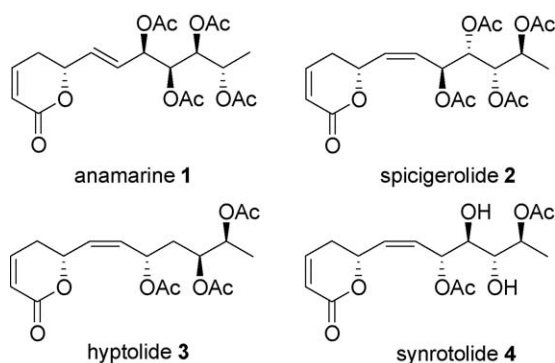
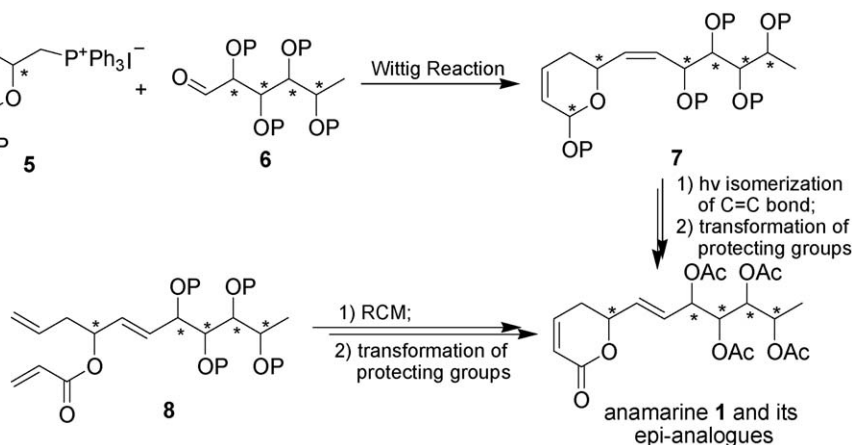
14]. Due to inconvenience of *h* $\nu$  irradiation reaction and expensive Grubbs catalyst for RCM reaction, in our opinion new synthetic method for anamarine and its analogues should be developed. Additionally, structure-activity relationship (SAR) has demonstrated that the  $\alpha,\beta$ -unsaturated- $\delta$ -lactone scaffold of anamarine and its *epi*-isomers played a key role for their bioactivities because such structure unit was an excellent potential Michael acceptor for nucleophilic amino acid residues of the natural receptors [15–17]. In view of the similarity in size between fluorine atom and hydrogen atom and the strong electron-withdrawing property of *gem*-difluoromethylene group (CF<sub>2</sub>) [18,19], we intended to introduce a CF<sub>2</sub> group to  $\alpha,\beta$ -unsaturated- $\delta$ -lactone of anamarine at the  $\gamma$ -position. We envisioned that the resultant  $\gamma,\gamma$ -difluoromethylenated- $\alpha,\beta$ -unsaturated anamarine analogues would be much more electron deficient, making it a better candidate to enhance the reactivity of the conjugated double bond as an acceptor with minimum steric change. Herein we would like to describe the total synthesis of *gem*-difluoromethylenated analogues of anamarine using novel synthetic strategy.

## 2. Results and discussion

Our retrosynthetic analysis was outlined in Scheme 2. We proposed that  $\gamma,\gamma$ -*gem*-difluoromethylenated anamarine analogues **9–10** could be afforded from the intermediate **A** by means of oxidation–cyclization procedure, which was developed by Forsyth group [20] and have been successfully utilized to synthesize a series of *gem*-difluoromethylenated  $\alpha,\beta$ -unsaturated- $\delta$ -lactone derivatives in our group [21,22]. Homopropargylation of aldehyde **C** with fluorine-containing building block **B** would yield compound **A**. Aldehyde **C** would be prepared via reduction of ester **D** and

\* Corresponding author at: Shanghai Institute of Organic Chemistry, Key Laboratory of Organofluorine Chemistry, Chinese Academy of Sciences, 354 Fenglin Lu, Shanghai 200032, China. Fax: +86 21 64166128.

E-mail address: [flq@mail.sioc.ac.cn](mailto:flq@mail.sioc.ac.cn) (F.-L. Qing).



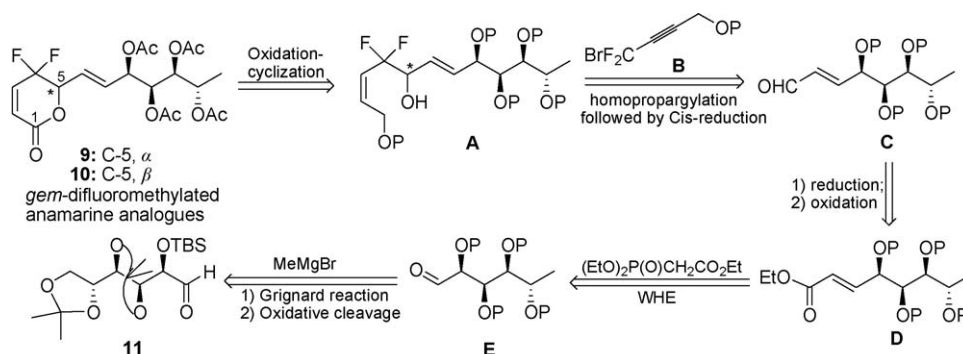
subsequent oxidation. The ester **D** could be obtained by means of WHE reaction between  $(\text{EtO})_2\text{P}(\text{O})\text{CH}_2\text{CO}_2\text{Et}$  and aldehyde **E**, which, in turn, would be produced from readily prepared aldehyde **11** via Grignard reaction followed by oxidative cleavage.

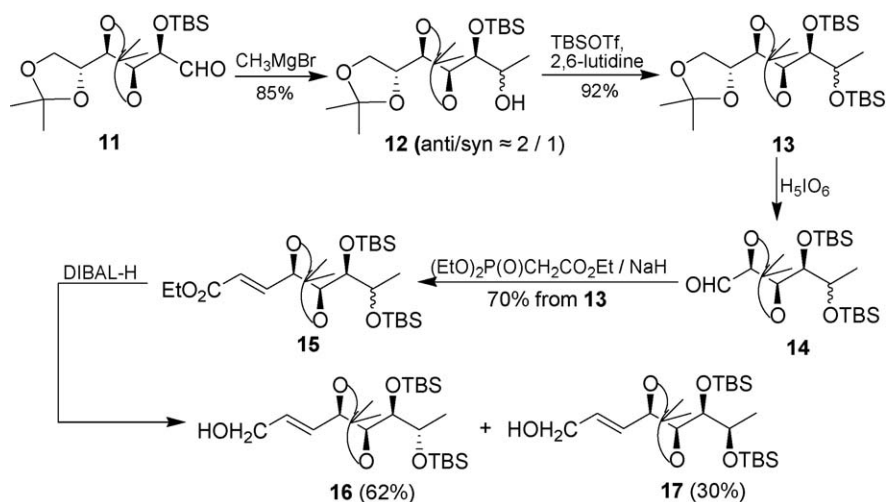
Based on our retrosynthetic analysis, our synthesis embarked from the aldehyde **11** (Scheme 3), which was prepared from commercially available  $\text{D}$ -glucono- $\delta$ -lactone in 3 steps according to reported procedure [23,24]. The nucleophilic addition of aldehyde **11** with  $\text{MeMgBr}$  gave alcohol **12** as a mixture of two diastereomers with *anti*-isomer as the major product [25] and two diastereoisomers could not be separated by flash chromatography. Silylation of alcohol **12** with  $\text{TBSOTf}/2,6$ -lutidine produced the compound **13** in good yield. When **13** was treated with periodic acid hydrate ( $\text{H}_5\text{IO}_6$ ) [26], the selective hydrolysis of terminal isopropylidene acetal in **13** and *in situ* glycol cleavage proceeded smoothly in one

pot to give aldehyde **14**, which was used in next step without purification. Then, treatment of **14** with ethyl 2-(diethoxyphosphoryl)acetate in the presence of  $\text{NaH}$  afforded ester **15** in 70% yields over two steps. Reduction of ester **15** with  $\text{DIBAL-H}$  smoothly provided the alcohols **16** and **17** in good yield. Fortunately, two diastereomers **16** and **17** could be readily separated by silica gel chromatography.

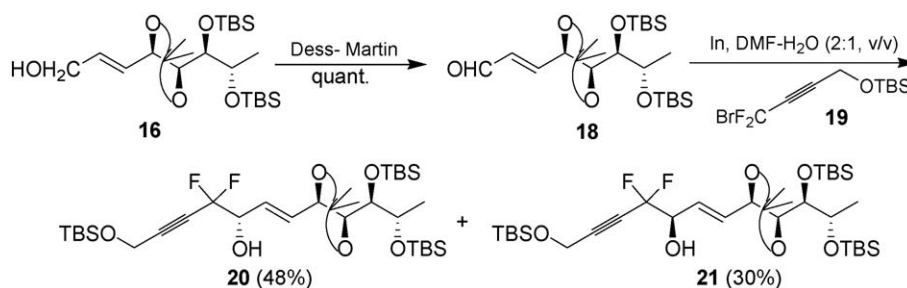
Oxidation of the alcohol **16** with Dess-Martin oxidant gave the desired aldehyde **18** in almost quantitative yield (Scheme 4). At this point, we focused our efforts on the homopropargylation of aldehyde **18** with fluorine-containing building block **19** [21]. Initially, the reaction conditions developed by Hammond group [27] were used. However, we found that treatment of aldehyde **18** with (4-bromo-4,4-difluorobut-2-ynoxy)(*tert*-butyl)dimethylsilane **19** with  $\text{THF-H}_2\text{O}$  (1:4, v/v) as solvent in the presence of indium at room temperature gave the expected product **20** in low yield along with its diastereomer **21** (*anti/syn* = 1.6:1, determined by  $^{19}\text{F}$  NMR before column chromatography). Diastereoisomer **20** and **21** could be separated by silica gel chromatography. After further optimization, we were pleased to find that substitution of  $\text{DMF-H}_2\text{O}$  (2:1, v/v) for  $\text{THF-H}_2\text{O}$  (1:4, v/v) as solvent would significantly improve the yield of alcohols **20-21** and *anti*-alcohol **20** was still the major diastereoisomer [28,29].

With the key intermediates **20-21** in hand, the synthesis of the target molecules were performed as outlined in Scheme 5. Initial attempts to hydrogenate the triple bond of in **20** and **21** to the *cis* double bond using Lindlar catalyst/quinoline system failed. Fortunately, the selective hydrogenation progressed well utilizing  $\text{Pd-BaSO}_4$ -quinoline system [30] and compound **22** and **23** was provided in 96% and 93% yields, respectively. Selective deprotection of the primary TBS group in **22** and **23** with  $\text{D}$ -camphor-10-





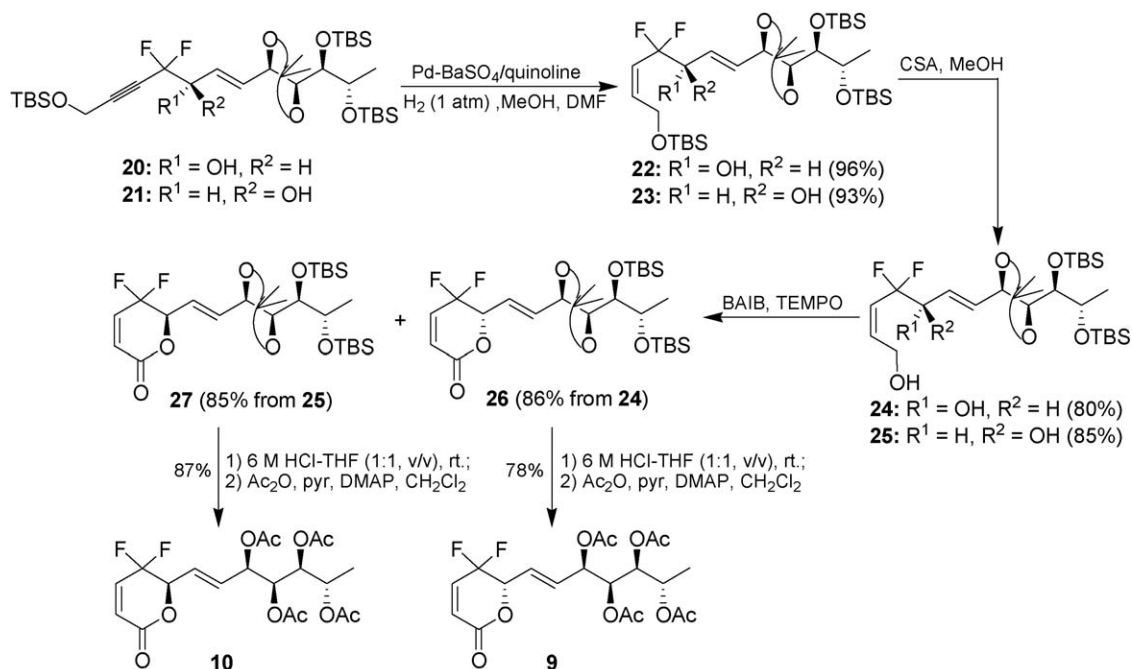
Scheme 3.



Scheme 4.

sulfonic acid (CSA) gave cyclization precursor **24** and **25** in 80% and 85% yield, respectively. Delightfully, treatment of compound **24** and **25** with 0.2 equiv. of 2,2,6,6-tetramethyl-1-piperidinyloxy (TEMPO)/3.0 equiv. of [bis(acetoxy)iodo]benzene (BAIB) in dichloromethane at room temperature smoothly afforded cycliza-

tion compound **26** and **27** in 86% and 85% yield, respectively. Using O'Doherty reported reaction condition [14], attempts to removal of all protecting groups by heating the lactones **26** and **27** in 10% aqueous HCl/THF for 10 min at 65 °C failed. Gratifyingly, treatment of compounds **26** and **27** with 6 M aqueous HCl/THF (1:1, v/v)



Scheme 5.

followed by direct acetylation with Ac<sub>2</sub>O/DMAP/pyridine successfully afforded out target molecules **9** and **10** in 78% and 87% yields, respectively.

### 3. Conclusions

We have accomplished the total synthesis of *gem*-difluoromethylenated analogues of anamarine **9** and **10** in a straightforward fashion. Our synthesis featured practical preparation of the key intermediates **20–21** through the indium-mediated *gem*-difluoropropargylation of aldehyde **18** with the fluorine-containing building block **19** and efficient construction of  $\alpha,\beta$ -unsaturated- $\delta$ -lactone scaffold via BAIB/TEMPO procedure. In our opinion, herein reported synthetic route provided a novel optional method for the preparation of anamarine and its analogues.

### 4. Experimental

#### 4.1. General

Unless otherwise indicated, all chemicals and solvents were used as received from commercial sources or purified by standard procedures. Optical rotations were recorded on a Jasco P-1030 polarimeter. IR Spectra were scanned with a Bio-Rad FTS185 spectrophotometer. <sup>1</sup>H- and <sup>13</sup>C NMR spectra were obtained using a Bruker AM300 and AM400 spectrometer, respectively and <sup>19</sup>F NMR spectra were recorded on a Bruker AM300 spectrometer (CFCl<sub>3</sub> as external standard and low field is positive. Chemical shifts ( $\delta$ ) in ppm, coupling constants (*J*) in Hz). LRMS were measured on Agilent system mass spectrometer and HRMS on an APEXIII (7.0 T) FTMS or waters mass spectrometer, respectively. Elemental analyses were taken on a Vario EL III elementary analysis instrument.

#### 4.2. (1S)-1-(*tert*-Butyldimethylsilyloxy)-1-((4S,5R)-5-((R)-2,2-dimethyl-1,3-dioxolan-4-yl)-2,2-dimethyl-1,3-dioxolan-4-yl)propan-2-ol (**12**)

To a solution of compound **11** (9.60 g, 26.7 mmol) in Et<sub>2</sub>O (100 mL) at  $-78^{\circ}\text{C}$  was added MeMgBr (17.8 mL, 3 M solution in ether) dropwise. After stirring for 1 h, the reaction was quenched with saturated aqueous NH<sub>4</sub>Cl and was extracted with EtOAc. The combined organic phases were washed with brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Removal of all the solvent in vacuo resulted in a residue, which was purified by silica gel chromatography (petroleum ether:ethyl acetate = 4:1) to give compound **12** (8.85 g, 85% yield) as a colorless oil: <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  4.14 (m, 2H), 4.02 (m, 2H), 3.90 (m, 2H), 3.71 (m, 0.6H), 3.60 (m, 0.4H), 2.29 (br s, 1H), 1.41 (m, 6H), 1.35 (m, 6H), 1.24 (m, 3H), 0.94 (s, 9H), 0.13 (m, 6H); IR (thin film)  $\nu_{\text{max}}$  3492, 2932, 1474, 1254, 1070, 838 cm<sup>-1</sup>; MS (ESI) *m/z* 391 (M+H)<sup>+</sup>, 413 (M+Na)<sup>+</sup>; Anal. Calcd. for C<sub>19</sub>H<sub>38</sub>O<sub>6</sub>Si: C, 58.43; H, 9.81. Found: C, 58.53; H, 9.65.

#### 4.3. (R)-4-((4R,5S)-5-((1R)-1,2-bis(*tert*-Butyldimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)-2,2-dimethyl-1,3-dioxolane (**13**)

To a solution of compound **12** (6.50 g, 16.7 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (100 mL) was added 2,6-lutidine (4.3 mL, 45.0 mmol) followed by TBSOTf (6.5 mL, 33.0 mmol) dropwise at  $-10^{\circ}\text{C}$ . The reaction mixture was stirred at  $-10^{\circ}\text{C}$  for 0.5 h, then warmed to room temperature and stirred overnight. The reaction mixture was quenched with saturated aqueous NaHCO<sub>3</sub> and was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic extracts were washed with 1 M HCl and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Removal of all the solvent in vacuo gave a residue, which was purified by silica gel

chromatography (petroleum ether:ethyl acetate = 20:1) to afford compound **13** (7.74 g, 92% yield) as a clear oil: <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  4.22–4.02 (m, 3H), 3.99–3.74 (m, 3H), 3.73 (dd, *J* = 4.5 Hz, 1.2 Hz, 0.4H), 3.69 (t, *J* = 3.9 Hz, 0.6H), 1.37 (m, 6H), 1.33 (m, 6H), 1.19 (t, *J* = 6.6 Hz, 3H), 0.91 (m, 18H), 0.09 (m, 12H); IR (thin film)  $\nu_{\text{max}}$  2933, 1473, 1255, 835 cm<sup>-1</sup>; MS (ESI) *m/z* 505 (M+H)<sup>+</sup>; Anal. Calcd. for C<sub>25</sub>H<sub>52</sub>O<sub>6</sub>Si<sub>2</sub>: C, 59.48; H, 10.38. Found: C, 59.64; H, 10.39.

#### 4.4. (E)-Ethyl 3-((4R,5S)-5-((1R)-1,2-bis(*tert*-butyldimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)acrylate (**15**)

To a solution of H<sub>5</sub>IO<sub>6</sub> (4.40 g, 19.3 mmol) in Et<sub>2</sub>O (80 mL) was added a solution of compound **13** (6.50 g, 12.9 mmol) in Et<sub>2</sub>O (40 mL) dropwise at room temperature. After stirring for 20 h the reaction mixture was filtered through Celite and the filter cake was washed with Et<sub>2</sub>O. The filtrate was added H<sub>2</sub>O (50 mL) and the organic phase was separated. The aqueous layer was extracted with Et<sub>2</sub>O and the combined organic extracts were washed with H<sub>2</sub>O and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Removal of all the solvent in vacuo gave a residue, which was used without further purification.

To a solution of (EtO)<sub>2</sub>P(O)CH<sub>2</sub>CO<sub>2</sub>Et (4.33 g, 19.4 mmol) in THF (80 mL) was added NaH (774 mg, 60% in oil, 19.4 mmol) at 0 °C. The reaction mixture was stirred for 30 min, then a solution of the above residue in THF (30 mL) was added dropwise. After stirring for 1 h, the reaction was quenched with saturated aqueous NH<sub>4</sub>Cl and was extracted with Et<sub>2</sub>O. The combined organic phases were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. After filtration and removal of all the solvent, the residue was purified by silica gel chromatography (petroleum ether:ethyl acetate = 100:1) to give compound **15** (4.53 g, 70% yield two steps) as a colorless oil: <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.14 (dd, *J* = 15.6 Hz, 4.5 Hz, 0.3H), 6.88 (dd, *J* = 15.6 Hz, 5.4 Hz, 0.7H), 6.13 (d, *J* = 15.6 Hz, 1H), 4.51 (m, 1H), 4.20 (m, 2H), 3.88 (m, 2H), 3.62 (t, *J* = 3.9 Hz, 0.7H), 3.58 (t, *J* = 4.8 Hz, 0.3H), 1.44 (s, 3H), 1.39 (s, 3H), 1.28 (m, 3H), 1.17 (d, *J* = 6.0 Hz, 3H), 0.91 (m, 18H), 0.09 (m, 12H); IR (thin film)  $\nu_{\text{max}}$  2933, 1728, 1662, 1473, 1257, 1104, 836 cm<sup>-1</sup>; MS (ESI) *m/z* 525 (M+Na)<sup>+</sup>, 541 (M+K)<sup>+</sup>; Anal. Calcd. for C<sub>25</sub>H<sub>50</sub>O<sub>6</sub>Si<sub>2</sub>: C, 59.72; H, 10.02. Found: C, 59.49; H, 10.24.

#### 4.5. (E)-3-((4R,5S)-5-((1R,2S)-1,2-bis(*tert*-Butyldimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)prop-2-en-1-ol (**16**) and (E)-3-((4R,5S)-5-((1R,2R)-1,2-bis(*tert*-butyldimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)prop-2-en-1-ol (**17**)

To a solution of **15** (4.20 g, 8.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (80 mL) at 0 °C was added DIBAL-H (12.5 mL, 1 M solution in toluene, 12.5 mmol) dropwise. After stirring for 1 h, the reaction was quenched with saturated aqueous Rochelle's salt. Warmed up to room temperature, the mixture was stirred for 3 h. The aqueous phase was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic phase was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. After filtration and removal of all the solvent in vacuo, the residue was purified by silica gel chromatography (petroleum ether:ethyl acetate = 20:1) to afford compound **16** (2.39 g, 62% yield) and **17** (1.16 g, 30% yield). Compound **16**: clear oil; [ $\alpha$ ]<sub>D</sub><sup>27</sup> = +13.6° (*c* 1.75, CHCl<sub>3</sub>); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  5.98 (dt, *J* = 15.6 Hz, 5.4 Hz, 1H), 5.70 (dd, *J* = 15.3 Hz, 7.5 Hz, 1H), 4.30 (t, *J* = 8.4 Hz, 1H), 4.17 (d, *J* = 4.8 Hz, 2H), 3.82 (m, 1H), 3.75 (dd, *J* = 9.0 Hz, 3.6 Hz, 1H), 3.58 (t, *J* = 3.6 Hz, 1H), 1.41 (s, 3H), 1.39 (s, 3H), 1.12 (d, *J* = 6.3 Hz, 3H), 0.93 (s, 9H), 0.88 (s, 9H), 0.12 (s, 3H), 0.09 (s, 3H), 0.05 (s, 6H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  134.6, 128.1, 108.6, 81.7, 77.5, 76.7, 69.7, 62.5, 27.0, 26.0, 25.9, 18.8, 18.5, 18.1, -3.9, -4.3, -4.4, -4.8; IR (thin film)  $\nu_{\text{max}}$  3420, 2931, 1473, 1254, 1101, 835 cm<sup>-1</sup>; MS (ESI) *m/z* 483 (M+Na)<sup>+</sup>; HRMS Calcd. for

$C_{23}H_{48}O_5Si_2Na$ : 483.2933; found: 483.2939. Compound **17**: clear oil;  $[\alpha]_D^{25} = +24.4^\circ$  (c 1.00,  $CHCl_3$ );  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  5.97 (dt,  $J = 15.3$  Hz, 5.4 Hz, 1H), 5.79 (dd,  $J = 15.3$  Hz, 6.9 Hz, 1H), 4.37 (t,  $J = 7.8$  Hz, 1H), 4.16 (d,  $J = 5.1$  Hz, 2H), 3.92 (dd,  $J = 9.0$  Hz, 3.6 Hz, 1H), 3.83 (m, 1H), 3.50 (t,  $J = 3.6$  Hz, 1H), 1.41 (s, 6H), 1.17 (d,  $J = 6.3$  Hz, 3H), 0.91 (s, 9H), 0.87 (s, 9H), 0.09 (s, 3H), 0.07 (s, 3H), 0.06 (s, 3H), 0.03 (s, 3H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  133.3, 129.2, 108.6, 79.7, 77.5, 73.8, 70.6, 62.9, 27.1, 27.0, 25.8, 18.1, -4.4, -4.9; IR (thin film)  $\nu_{max}$  3413, 2858, 1473, 1254, 1102, 835  $cm^{-1}$ ; MS (ESI)  $m/z$  483 (M+Na) $^+$ ; HRMS Calcd. for  $C_{23}H_{48}O_5Si_2Na$ : 483.2933; found: 483.2937.

4.6. (S,E)-1-((4R,5S)-5-((1R,2S)-1,2-bis(tert-butylidimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)-7-(tert-butylidimethylsilyloxy)-4,4-difluorohept-1-en-5-yn-3-ol (**20**) and (R,E)-1-((4R,5S)-5-((1R,2S)-1,2-bis(tert-butylidimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)-7-(tert-butylidimethylsilyloxy)-4,4-difluorohept-1-en-5-yn-3-ol (**21**)

To a solution of **16** (0.74 g, 1.61 mmol) in  $CH_2Cl_2$  (20 mL) was added Dess-Martin reagent (1.02 g, 2.41 mmol). After stirring for 1 h, the reaction mixture was quenched with saturated aqueous  $Na_2S_2O_3$  and was extracted with  $CH_2Cl_2$ . The combined organic extracts were washed with brine, dried over anhydrous  $Na_2SO_4$ . After filtration and removal of all the solvent, the residue was used for next step without further purification.

To a stirred suspension of the above residue and compound **19** (0.48 g, 1.39 mmol) in DMF- $H_2O$  (20 mL, 2:1, v/v) was added indium power (0.20 g, 1.77 mmol) at room temperature. After stirring for 5 h, the reaction mixture was quenched with 1 M HCl. The aqueous phase was extracted with EtOAc, and the organic phase was washed with water and brine, dried over anhydrous  $Na_2SO_4$ . After filtration and removal of all the solvent, the residue was purified by silica gel chromatography (petroleum ether:ethyl acetate = 20:1) to afford compound **20** (0.52 g, 48% yield) and **21** (0.33 g, 30% yield). Compound **20**: yellow oil;  $[\alpha]_D^{24} = +1.2^\circ$  (c 0.53,  $CHCl_3$ );  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  5.94 (m, 2H), 4.36 (m, 4H), 3.78 (m, 2H), 3.59 (t,  $J = 3.9$  Hz, 1H), 2.02 (br s, 1H), 1.42 (s, 3H), 1.38 (s, 3H), 1.12 (d,  $J = 6.0$  Hz, 3H), 0.90 (m, 27H), 0.09 (m, 18H);  $^{19}F$  NMR (282 MHz,  $CDCl_3$ )  $\delta$  -95.1 (d,  $J = 286.1$  Hz, 1F), -96.6 (d,  $J = 272.4$  Hz, 1F); IR (thin film)  $\nu_{max}$  3460, 2932, 2230, 1473, 1256, 1104, 836  $cm^{-1}$ ; MS (ESI)  $m/z$  701 (M+Na) $^+$ , 717 (M+K) $^+$ ; Anal. Calcd. for  $C_{33}H_{64}F_2O_6Si_3$ : C, 58.36; H, 9.50. Found: C, 58.68; H, 9.24. Compound **21**: yellow oil;  $[\alpha]_D^{25} = +25.2^\circ$  (c 2.00,  $CHCl_3$ );  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  5.90 (m, 2H), 4.36 (m, 4H), 3.81 (m, 1H), 3.75 (dd,  $J = 8.4$  Hz, 4.5 Hz, 1H), 3.60 (t,  $J = 3.6$  Hz, 1H), 1.80 (br s, 1H), 1.42 (s, 3H), 1.38 (s, 3H), 1.12 (d,  $J = 6.3$  Hz, 3H), 0.91 (m, 27H), 0.09 (m, 18H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  134.0, 127.7, 120.9 (t,  $J = 237.3$  Hz), 109.0, 88.2 (t,  $J = 6.4$  Hz), 81.8, 77.0, 76.7, 75.5 (t,  $J = 38.9$  Hz), 74.4 (t,  $J = 29.8$  Hz), 69.8, 51.2, 27.0, 26.9, 26.1, 25.9, 25.7, 18.7, 18.5, 18.2, 18.1, -3.9, -4.4, -4.5, -4.8, -5.3;  $^{19}F$  NMR (282 MHz,  $CDCl_3$ )  $\delta$  -94.5 (d,  $J = 274.4$  Hz, 1F), -96.6 (d,  $J = 280.7$  Hz, 1F); IR (thin film)  $\nu_{max}$  3460, 2932, 2258, 1650, 1473, 1257, 1108, 836  $cm^{-1}$ ; MS (ESI)  $m/z$  696 (M+ $NH_4$ ) $^+$ ; HRMS Calcd. for  $C_{33}H_{64}F_2O_6Si_3Na$ : 701.3871; found: 701.3838.

4.7. (S,1E,5Z)-1-((4R,5S)-5-((1R,2S)-1,2-bis(tert-butylidimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)-7-(tert-butylidimethylsilyloxy)-4,4-difluorohepta-1,5-dien-3-ol (**22**)

To a mixture of Pd-BaSO<sub>4</sub> (28 mg, 42 mg/mmol) in MeOH (20 mL) was added a solution of quinoline (28 mg, 42 mg/mmol) in MeOH (2 mL) at 0 °C. Warming up to room temperature, the suspension mixture was stirred for 15 min before a solution of compound **20** (450 mg, 0.66 mmol) in DMF (10 mL) was added. After the mixture was stirred for 3.5 h under hydrogen atmosphere

(1 atm) at 35 °C,  $^{19}F$  NMR indicated the absence of starting material **20**. The reaction mixture was filtered and concentrated. The residue was purified by silica gel chromatography (petroleum ether:ethyl acetate = 20:1) to afford compound **22** (449 mg, 96% yield) as a yellow oil;  $[\alpha]_D^{24} = +4.8^\circ$  (c 0.60,  $CHCl_3$ );  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  6.00 (m, 1H), 5.90 (m, 2H), 5.47 (m, 1H), 4.44 (m, 2H), 4.33 (m, 2H), 3.80 (m, 1H), 3.75 (dd,  $J = 8.1$  Hz, 3.9 Hz, 1H), 3.58 (t,  $J = 3.6$  Hz, 1H), 2.10 (br s, 1H), 1.42 (s, 3H), 1.38 (s, 3H), 1.12 (d,  $J = 6.6$  Hz, 3H), 0.90 (m, 27H), 0.07 (m, 18H);  $^{19}F$  NMR (282 MHz,  $CDCl_3$ )  $\delta$  -101.6 (d,  $J = 260.9$  Hz, 1F), -104.4 (dt,  $J = 261.5$  Hz, 13.2 Hz, 1F); IR (thin film)  $\nu_{max}$  3460, 2932, 1473, 1256, 1104, 836  $cm^{-1}$ ; MS (ESI)  $m/z$  703 (M+Na) $^+$ , 719 (M+K) $^+$ ; Anal. Calcd. for  $C_{33}H_{66}F_2O_6Si_3$ : C, 58.19; H, 9.77. Found: C, 57.95; H, 9.72.

4.8. (R,1E,5Z)-1-((4R,5S)-5-((1R,2R)-1,2-bis(tert-butylidimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)-7-(tert-butylidimethylsilyloxy)-4,4-difluorohepta-1,5-dien-3-ol (**23**)

Compound **23** was prepared from compound **21** (300 mg, 0.44 mmol) in 93% yield using the same conditions as described for compound **22**. Yellow oil;  $[\alpha]_D^{24} = +20.0^\circ$  (c 1.50,  $CHCl_3$ );  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  5.99 (m, 1H), 5.86 (m, 2H), 5.48 (m, 1H), 4.44 (m, 2H), 4.30 (m, 2H), 3.80 (m, 1H), 3.74 (dd,  $J = 8.1$  Hz, 3.9 Hz, 1H), 3.58 (t,  $J = 3.3$  Hz, 1H), 2.29 (br s, 1H), 1.42 (s, 3H), 1.38 (s, 3H), 1.12 (d,  $J = 6.6$  Hz, 3H), 0.90 (m, 27H), 0.08 (m, 18H);  $^{19}F$  NMR (282 MHz,  $CDCl_3$ )  $\delta$  -101.6 (d,  $J = 274.1$  Hz, 1F), -104.9 (dt,  $J = 259.7$  Hz, 12.6 Hz, 1F); IR (thin film)  $\nu_{max}$  3480, 2932, 1650, 1473, 1256, 1104, 836  $cm^{-1}$ ; MS (ESI)  $m/z$  703 (M+Na) $^+$ , 719 (M+K) $^+$ ; Anal. Calcd. for  $C_{33}H_{66}F_2O_6Si_3$ : C, 58.19; H, 9.77. Found: C, 58.53; H, 9.54.

4.9. (S,2Z,6E)-7-((4R,5S)-5-((1R,2R)-1,2-bis(tert-butylidimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)-4,4-difluorohepta-2,6-diene-1,5-diol (**24**)

To a solution of **22** (400 mg, 0.59 mmol) in MeOH (10 mL) was added 10-CSA (9 mg, 0.029 mmol). After stirring for 1 h, the reaction mixture was quenched with saturated  $NaHCO_3$ . The aqueous phase was extracted with EtOAc, and the organic phase was washed with brine, dried over anhydrous  $Na_2SO_4$ . After filtration and removal of all the solvent in vacuo, the residue was purified by silica gel chromatography (petroleum ether:ethyl acetate = 4:1) to give compound **24** (267 mg, 80% yield) as a clear oil;  $[\alpha]_D^{23} = +7.3^\circ$  (c 1.00,  $CHCl_3$ );  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  6.03 (m, 1H), 5.88 (m, 2H), 5.51 (m, 1H), 4.36 (m, 4H), 3.79 (m, 2H), 3.57 (t,  $J = 3.3$  Hz, 1H), 2.62 (br s, 2H), 1.42 (s, 3H), 1.39 (s, 3H), 1.13 (d,  $J = 6.6$  Hz, 3H), 0.92 (s, 9H), 0.88 (s, 9H), 0.12 (s, 3H), 0.08 (s, 3H), 0.05 (s, 6H);  $^{13}C$  NMR (100 MHz,  $CDCl_3$ )  $\delta$  139.0 (t,  $J = 4.6$  Hz), 131.9, 129.1, 121.7 (t,  $J = 26.3$  Hz), 120.4 (dd,  $J = 244.8$  Hz, 243.4 Hz), 109.0, 81.8, 77.0, 76.6, 73.0 (t,  $J = 30.1$  Hz), 69.8, 58.6, 27.0, 26.8, 26.1, 25.9, 18.9, 18.5, 18.1, -3.9, -4.4, -4.8;  $^{19}F$  NMR (282 MHz,  $CDCl_3$ )  $\delta$  -101.1 (dt,  $J = 270.1$  Hz, 13.5 Hz, 1F), -102.5 (dt,  $J = 258.0$  Hz, 12.9 Hz, 1F); IR (thin film)  $\nu_{max}$  3450, 2956, 1473, 1255, 1109, 836  $cm^{-1}$ ; MS (ESI)  $m/z$  611 (M+COOH) $^-$ ; HRMS Calcd. for  $C_{28}H_{53}F_2O_8Si_2$ : 611.3253; found: 611.3262.

4.10. (R,2Z,6E)-7-((4R,5S)-5-((1R,2R)-1,2-bis(tert-butylidimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)-4,4-difluorohepta-2,6-diene-1,5-diol (**25**)

Compound **25** was prepared from compound **23** (250 mg, 0.37 mmol) in 85% yield using the same conditions as described for compound **24**. Clear oil;  $[\alpha]_D^{26} = +26.2^\circ$  (c 1.00,  $CHCl_3$ );  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  6.05 (m, 1H), 5.88 (m, 2H), 5.52 (m, 1H), 4.35 (m, 4H), 3.79 (m, 2H), 3.58 (t,  $J = 4.5$  Hz, 1H), 2.05 (br s, 2H), 1.42 (s, 3H), 1.39 (s, 3H), 1.13 (d,  $J = 5.7$  Hz, 3H), 0.92 (s, 9H), 0.88 (s, 9H), 0.12 (s, 3H), 0.09 (s, 3H), 0.06 (s, 6H).  $^{19}F$  NMR (282 MHz,  $CDCl_3$ )  $\delta$

–99.5 (dt,  $J = 259.7$  Hz, 9.8 Hz, 1F), –103.1 (d,  $J = 260.3$  Hz, 1F); IR (thin film)  $\nu_{\max}$  3450, 2933, 1469, 1379, 1107, 834  $\text{cm}^{-1}$ ; MS (ESI)  $m/z$  611 ( $\text{M}+\text{COOH}^-$ ); Anal. Calcd. for  $\text{C}_{27}\text{H}_{52}\text{F}_2\text{O}_6\text{Si}_2$ : C, 57.21; H, 9.25. Found: C, 57.42; H, 9.25.

4.11. (*S*)-6-((*E*)-2-((4*R*,5*S*)-5-((1*R*,2*R*)-1,2-bis(*tert*-Butyldimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)vinyl)-5,5-difluoro-5,6-dihydropyran-2-one (26)

To a solution of **24** (220 mg, 0.39 mmol) in  $\text{CH}_2\text{Cl}_2$  (5 mL) was added BAIB (520 mg, 1.17 mmol) and TEMPO (12 mg, 20 mmol%) at room temperature. After stirring for 3 h, the reaction mixture was quenched with saturated aqueous  $\text{Na}_2\text{S}_2\text{O}_3$  and was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic extracts were washed with saturated  $\text{NaHCO}_3$  and brine; dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After filtration and removal of all the solvent in vacuo, the residue was purified by silica gel chromatography (petroleum ether:ethyl acetate = 10:1) to afford compound **26** (188 mg, 86% yield) as a clear oil:  $[\alpha]_{\text{D}}^{24} = +41.8^\circ$  ( $c$  1.20,  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  6.82 (m, 1H), 6.32 (d,  $J = 9.9$  Hz, 1H), 6.08 (dd,  $J = 15.3$  Hz, 6.0 Hz, 1H), 5.94 (dd,  $J = 15.6$  Hz, 5.4 Hz, 1H), 5.06 (dt,  $J = 15.6$  Hz, 6.6 Hz, 1H), 4.40 (t,  $J = 6.6$  Hz, 1H), 3.80 (m, 2H), 3.60 (t,  $J = 3.3$  Hz, 1H), 1.43 (s, 3H), 1.39 (s, 3H), 1.13 (d,  $J = 6.0$  Hz, 3H), 0.92 (s, 9H), 0.88 (s, 9H), 0.13 (s, 3H), 0.08 (s, 3H), 0.05 (s, 6H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  159.9 (t,  $J = 2.0$  Hz), 137.6 (dd,  $J = 31.3$  Hz, 26.8 Hz), 135.8, 126.6 (dd,  $J = 9.8$  Hz, 8.3 Hz), 121.7 (dd,  $J = 3.2$  Hz, 1.0 Hz), 111.8 (dd,  $J = 243.9$  Hz, 238.1 Hz), 109.2, 81.7, 78.7 (dd,  $J = 32.7$  Hz, 28.6 Hz), 76.6, 76.5, 69.8, 27.0, 26.8, 26.0, 25.9, 19.0, 18.5, 18.1, –3.9, –4.4, –4.8;  $^{19}\text{F}$  NMR (282 MHz,  $\text{CDCl}_3$ )  $\delta$  –107.1 (dd,  $J = 293.9$  Hz, 15.2 Hz, 1F), –108.7 (dd,  $J = 293.3$  Hz, 8.9 Hz, 1F); IR (thin film)  $\nu_{\max}$  2932, 1754, 1615, 1473, 1383, 1258, 1106, 835  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  (%) 505 (2), 303 (21), 159 (54), 131 (73), 73 (100); HRMS Calcd. for  $\text{C}_{23}\text{H}_{39}\text{F}_2\text{O}_6\text{Si}_2$ : 505.2253; found: 505.2249.

4.12. (*R*)-6-((*E*)-2-((4*R*,5*S*)-5-((1*R*,2*R*)-1,2-bis(*tert*-Butyldimethylsilyloxy)propyl)-2,2-dimethyl-1,3-dioxolan-4-yl)vinyl)-5,5-difluoro-5,6-dihydropyran-2-one (27)

Compound **27** was prepared from compound **25** (200 mg, 0.35 mmol) in 85% yield using the same conditions as described for compound **26**. White solid, mp 92–94 °C;  $[\alpha]_{\text{D}}^{25} = -3.7^\circ$  ( $c$  1.20,  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  6.82 (m, 1H), 6.32 (d,  $J = 10.5$  Hz, 1H), 6.04 (dd,  $J = 15.6$  Hz, 6.0 Hz, 1H), 5.93 (dd,  $J = 15.0$  Hz, 6.0 Hz, 1H), 5.02 (m, 1H), 4.40 (t,  $J = 6.6$  Hz, 1H), 3.82 (m, 1H), 3.77 (dd,  $J = 8.4$  Hz, 3.9 Hz, 1H), 3.61 (t,  $J = 3.3$  Hz, 1H), 1.43 (s, 3H), 1.39 (s, 3H), 1.13 (d,  $J = 6.3$  Hz, 3H), 0.92 (s, 9H), 0.88 (s, 9H), 0.13 (s, 3H), 0.09 (s, 3H), 0.06 (s, 6H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  159.9, 137.6 (dd,  $J = 31.2$  Hz, 26.5 Hz), 136.5, 126.5 (dd,  $J = 9.7$  Hz, 8.3 Hz), 121.6 (dd,  $J = 3.5$  Hz, 0.9 Hz), 111.7 (dd,  $J = 244.1$  Hz, 237.5 Hz), 109.2, 81.8, 79.1 (dd,  $J = 32.7$  Hz, 28.6 Hz), 76.6, 76.5, 69.8, 27.0, 26.8, 26.0, 25.9, 18.9, 18.4, 18.1, –3.9, –4.5, –4.8;  $^{19}\text{F}$  NMR (282 MHz,  $\text{CDCl}_3$ )  $\delta$  –107.1 (dd,  $J = 293.9$  Hz, 15.5 Hz, 1F), –108.7 (dd,  $J = 293.6$  Hz, 9.8 Hz, 1F); IR (thin film)  $\nu_{\max}$  2932, 1754, 1650, 1473, 1257, 1107, 835  $\text{cm}^{-1}$ ; MS (EI)  $m/z$  (%) 547 (2), 505 (4), 303 (32), 73 (100); HRMS Calcd. for  $\text{C}_{23}\text{H}_{39}\text{F}_2\text{O}_6\text{Si}_2$ : 505.2253; found: 505.2260.

4.13. (6*S*)-5,5-Difluoro-6-[(1*E*,3*R*,4*S*,5*S*,6*S*)-3,4,5,6-tetrakis(acetyloxy)-1-hepten-1-yl]-5,6-dihydro-2*H*-Pyran-2-one (9)

To a solution of **26** (80 mg, 0.14 mmol) in THF (3.5 mL) was added 6 M HCl (3.5 mL) at room temperature. After TLC demonstrated that starting material was completely consumed, the reaction was concentrated to give a residue, which was used without further purification.

To a solution of above residue in  $\text{CH}_2\text{Cl}_2$  (5 mL) was added dropwise pyridine (0.15 mL), DMAP (3 mg) followed by  $\text{Ac}_2\text{O}$  (0.1 mL) at 0 °C. The reaction mixture was allowed to warm to room temperature overnight. The reaction mixture was quenched with saturated aqueous  $\text{NaHCO}_3$  and was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic extracts were washed with 1 M HCl and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After filtration and removal of all the solvent, the residue was purified by silica gel chromatography (petroleum ether:ethyl acetate = 1:1) to afford compound **9** (50 mg, 78% yield) as a clear oil:  $[\alpha]_{\text{D}}^{24} = +0.3^\circ$  ( $c$  1.25,  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  6.83 (m, 1H), 6.32 (d,  $J = 10.2$  Hz, 1H), 6.01 (dd,  $J = 15.9$  Hz, 5.4 Hz, 1H), 5.87 (dd,  $J = 15.9$  Hz, 5.1 Hz, 1H), 5.48 (t,  $J = 5.7$  Hz, 1H), 5.30 (dd,  $J = 6.0$  Hz, 4.8 Hz, 1H), 5.23 (dd,  $J = 5.7$  Hz, 4.5 Hz, 1H), 5.04 (ddd,  $J = 16.8$  Hz, 8.1 Hz, 5.7 Hz, 1H), 4.91 (dt,  $J = 12.6$  Hz, 6.0 Hz, 1H), 2.15 (s, 3H), 2.12 (s, 3H), 2.07 (s, 3H), 2.02 (s, 3H), 1.21 (d,  $J = 6.3$  Hz, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  170.0, 169.8, 169.6, 159.8, 137.5 (t,  $J = 28.9$  Hz), 131.5, 126.5 (t,  $J = 0.9$  Hz), 123.7, 111.8 (t,  $J = 240.1$  Hz), 78.6 (t,  $J = 30.9$  Hz), 71.4, 70.9, 70.1, 67.4, 21.0, 20.8, 20.7, 20.5, 15.4;  $^{19}\text{F}$  NMR (282 MHz,  $\text{CDCl}_3$ )  $\delta$  –105.8 to –109.1 (m, 2F); IR (thin film)  $\nu_{\max}$  2929, 1753, 1648, 1459, 1375, 1215, 1068  $\text{cm}^{-1}$ ; MS (ESI)  $m/z$  480 ( $\text{M}+\text{NH}_4^+$ ), 485 ( $\text{M}+\text{Na}^+$ ); HRMS Calcd. for  $\text{C}_{20}\text{H}_{24}\text{F}_2\text{O}_{10}\text{Na}$ : 485.1230; found: 485.1243.

4.14. (6*R*)-5,5-Difluoro-6-[(1*E*,3*R*,4*S*,5*S*,6*S*)-3,4,5,6-tetrakis(acetyloxy)-1-hepten-1-yl]-5,6-dihydro-2*H*-Pyran-2-one (10)

Compound **10** (36 mg, 87%) was prepared from compound **27** (50 mg, 0.089 mmol) using the same conditions as described for compound **9**. Clear oil:  $[\alpha]_{\text{D}}^{24} = -54.6^\circ$  ( $c$  0.79,  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  6.83 (m, 1H), 6.32 (d,  $J = 9.6$  Hz, 1H), 6.04 (dd,  $J = 16.5$  Hz, 6.0 Hz, 1H), 5.85 (dd,  $J = 15.6$  Hz, 5.4 Hz, 1H), 5.47 (t,  $J = 6.3$  Hz, 1H), 5.30 (dd,  $J = 6.3$  Hz, 3.9 Hz, 1H), 5.23 (dd,  $J = 6.0$  Hz, 4.5 Hz, 1H), 5.04 (dt,  $J = 16.8$  Hz, 6.0 Hz, 1H), 4.92 (dt,  $J = 12.6$  Hz, 6.0 Hz, 1H), 2.13 (s, 6H), 2.07 (s, 3H), 2.03 (s, 3H), 1.21 (d,  $J = 6.6$  Hz, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  170.0, 169.9, 169.8, 169.6, 159.7, 137.6 (t,  $J = 25.9$  Hz), 131.5, 126.5 (dd,  $J = 11.5$  Hz, 8.2 Hz), 123.5 (t,  $J = 3.0$  Hz), 111.7 (dd,  $J = 244.3$  Hz, 237.3 Hz), 78.4 (dd,  $J = 32.9$  Hz, 28.1 Hz), 71.5, 71.0, 70.2, 67.3, 21.0, 20.8, 20.5, 15.5;  $^{19}\text{F}$  NMR (282 MHz,  $\text{CDCl}_3$ )  $\delta$  –107.2 (dd,  $J = 289.1$  Hz, 16.9 Hz, 1F), –109.4 (dt,  $J = 289.1$  Hz, 6.2 Hz, 1F); IR (thin film)  $\nu_{\max}$  2929, 1758, 1640, 1433, 1374, 1216, 1067  $\text{cm}^{-1}$ ; MS (ESI)  $m/z$  485 ( $\text{M}+\text{Na}^+$ ); HRMS Calcd. for  $\text{C}_{20}\text{H}_{24}\text{F}_2\text{O}_{10}\text{Na}$ : 485.1230; found: 485.1226.

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## References

- [1] A. Alemany, C. Mañquez, C. Pascual, S. Valverde, A. Perales, J. Fayos, M. Martínez-Ripoll, *Tetrahedron Lett.* 20 (1979) 3579–3582.
- [2] A. Alemany, C. Mañquez, C. Pascual, S. Valverde, M. Martínez-Ripoll, J. Fayos, A. Perales, *Tetrahedron Lett.* 20 (1979) 3583–3586.
- [3] S.A. Achmad, T. Høyer, A. Kjær, L. Makmur, R. Norrestam, *Acta Chem. Scand.* 41B (1987) 599–609.
- [4] M.T.D. Coleman, R.B. English, D.E.A. Rivett, *Phytochemistry* 26 (1987) 1497–1499.
- [5] R. Pereda-Miranda, L. Hernandez, M.J. Villavicencio, M. Novelo, P. Ibarra, H. Chai, J.M. Pezzuto, *J. Nat. Prod.* 56 (1993) 583–593.
- [6] R. Pereda-Miranda, R. Fragosó-Serrano, C.M. Cerda-García-Rojas, *Tetrahedron* 57 (2001) 47–53.
- [7] K. Lorenz, F.W. Lichtenthaler, *Tetrahedron Lett.* 28 (1987) 6437–6440.
- [8] S. Valverde, A. Herradon, B. Herradon, R.M. Babanal, M. Marrtin-Lomas, *Tetrahedron* 43 (1987) 3499–3504.
- [9] F.W. Lichtenthaler, K. Lorenz, W.-y. Ma, *Tetrahedron Lett.* 28 (1987) 47–50.
- [10] M. Abbas, PhD Thesis, Tübingen University, 2002.
- [11] E. Falomir, J. Murga, P. Ruiz, M. Carda, J.A. Marco, *J. Org. Chem.* 68 (2003) 5672–5676.



- [12] S. Diaz-Oltra, J. Murga, E. Falomir, M. Carda, J.A. Marco, *Tetrahedron* 40 (2004) 2979–2985.
- [13] D. Gao, G.A. O'Doherty, *Org. Lett.* 7 (2005) 1069–1072.
- [14] D. Gao, G.A. O'Doherty, *J. Org. Chem.* 70 (2005) 9932–9939.
- [15] P. Kumar, S.V. Naidu, *J. Org. Chem.* 71 (2006) 3935–3941.
- [16] A. de Fatima, L.K. Kohn, M.A. Antonio, J.E. de Carvalho, R.A. Pilli, *Bioorg. Med. Chem.* 14 (2006) 622–631.
- [17] S.B. Buck, C. Hardouin, S. Ichikawa, D.R. Soenen, C.M. Gauss, I. Hwang, M.R. Swingle, K.M. Bonness, R.E. Honkanen, D.L. Boger, *J. Am. Chem. Soc.* 125 (2003) 15694–15695.
- [18] R.D. Chambers, *Fluorine in Organic Chemistry*, Wiley, New York, 1973.
- [19] R.E. Banks, B.E. Smart, J.C. Tatlow, *Organofluorine Chemistry: Principles and Commercial Applications*, Plenum Press, New York, 1994.
- [20] T.M. Hansen, G.J. Florence, P. Lugo-Mas, J. Chen, J.N. Abrams, C.J. Forsyth, *Tetrahedron Lett.* 44 (2003) 57–59.
- [21] Z.-W. You, Z.-X. Jiang, B.-L. Wang, F.-L. Qing, *J. Org. Chem.* 71 (2006) 7261–7267.
- [22] J. Xu, X. Zhang, X.-L. Qiu, F.-L. Qing, *Synthesis* (2009) 602–608.
- [23] S.-G. Hu, T.-S. Hu, Y.-L. Wu, *Org. Biomol. Chem.* 2 (2004) 2305–2310.
- [24] D.D. Long, M.D. Smith, A. Martin, J.R. Wheatley, D.G. Watkin, M. Müller, G.W.J. Fleet, *J. Chem. Soc., Perkin Trans. 1* (2002) 1982–1998.
- [25] X. Li, M. Tanasova, C. Vasileiou, B. Borhan, *J. Am. Chem. Soc.* 130 (2008) 1885–1893.
- [26] W.-L. Wu, Y.-L. Wu, *J. Org. Chem.* 58 (1993) 3586–3588.
- [27] Z.G. Wang, G.B. Hammond, *J. Org. Chem.* 65 (2000) 6547–6552.
- [28] Y. Hanzawa, K. Inazawa, A. Kon, H. Aoki, Y. Kobayashi, *Tetrahedron Lett.* 28 (1987) 659–662.
- [29] H.L. Sham, D.A. Betebenner, N.E. Wideburg, D.J. Kempf, J.J. Plattner, D.W. Norbeck, *J. Fluorine Chem.* 73 (1995) 221–224.
- [30] Y. Nakamura, M. Okada, A. Sato, H. Horikawa, M. Koura, A. Saito, T. Taguchi, *Tetrahedron* 61 (2005) 5741–5753.